Tutorial II

Classical Mechanics

Course: PH211 + PH403

- Find the degree of freedom of a carbon-dioxide molecule (O=C=O) moving in 3-dimensions under each of the following "models".
 - (a) The C=O bonds are rigid, and $\angle O$ -C-O is rigid and is equal to 180°.
 - (b) The C=O bonds are harmonic (flexible), but $\angle O$ -C-O is rigid and is equal to 180°.
 - (c) The C=O bonds are rigid, but \angle O-C-O is flexible.
 - (d) The C=O bonds and $\angle O$ -C-O are flexible.
- 2. Find the number of degree's of freedom of a CH_4 molecule according to the following models,
 - (a) All C-H bonds and all H-C-H angles (\angle H-C-H) are stiff
 - (b) All C-H bonds are harmonic but all ∠H-C-H are stiff
 - (c) All C-H bonds are stiff but all \angle H-C-H vary
 - (d) All C-H bonds and all ∠H-C-H are harmonic
- 3. Use the D'Alembert's principle of virtual work to obtain the equation of motion of the following 2-D systems:
 - (a) a simple pendulum.
 - (b) a particle moving in a parabolic $(y = ax^2)$ wire under gravity.
 - (c) a particle sliding down a wedge of angle α .